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⑤ Platinum co-ordination compounds.

⑤ A platinum co-ordination compound linkable to a monoclonal antibody by a functional group which forms part of a moiety which stabilises the antibody against in vivo hydrolysis. Also the use of such compounds in the treatment of cancer.

- 1 -

PLATINUM CO-ORDINATION COMPOUNDS

This invention relates to novel platinum co-ordination  
5 compounds for the treatment of cancer and which are also inter  
alia linkable to monoclonal antibodies. It also provides  
complexes comprising the novel platinum co-ordination complexes  
and monoclonal antibodies, for use as site-specific or  
disease-specific chemotherapeutic agents.

10

The use of platinum co-ordination compounds, especially  
cisplatin (cis-diammine-dichloroplatinum II) and certain analogues  
thereof, in the chemotherapeutic treatment of cancer is now an  
established clinical technique, although efforts persist to find  
15 improved compounds. The problem with such compounds when

administered as a composition together with an inert carrier or diluent is that they are absorbed generally into the systemic circulation from where they have a toxic effect on normal cells and tissues as well as on the diseased cells and tissues which they are designed to treat. In practice, the maximum dose that can be administered is limited not by pharmaceutical effectiveness but by toxicity, with the result that the patient suffers unpleasant or even severe side-effects.

10 In an attempt to render platinum compounds specific for certain types of tumour cell, European patent specification 0099133 proposes platinum complexed anti-tumour immunoglobulins prepared by reacting platinum salts, particularly  $K_2PtCl_4$ , with anti-tumour reactive immunoglobulins in for example phosphate buffered saline. The toxicity of the resulting complex is said to be lower than that of cisplatin. However, despite the presence of the immunoglobulin (which is an antibody produced from a tumour-associated antigen), the complexes are believed to be relatively non-specific in practical usage because of poorly defined metal stoichiometry and because they are hydrolysed in vivo before they reach the target tumour site, thereby losing their activity.

It has also been proposed in general pharmacological terms to link known chemotherapeutic agents to monoclonal antibodies for the purposes of rendering the agent site- or

disease-specific but again the problem of in vivo stability remains.

It is an object of the present invention to provide  
5 co-ordination compounds of platinum which inter alia are  
chemically linkable to monoclonal antibodies in such a way that  
the desired in vivo stability is obtained. It is a further  
object of the invention to provide conjugate platinum  
co-ordination compound/monoclonal antibody complexes for localised  
10 pharmacological activity and which are stable in vivo until  
they reach the target site.

We have found that platinum co-ordination compounds can  
be linked to monoclonal antibodies in such a way that the desired  
15 in vivo stability is achieved by providing the compounds with a  
linkable functional group which forms part of an  
antibody-stabilising moiety.

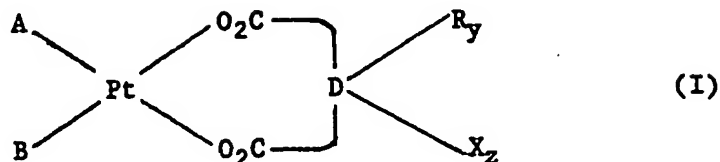
Accordingly, the present invention provides a  
20 co-ordination compound of platinum linkable to a monoclonal  
antibody, wherein the compound includes an antibody-linkable  
functional group which forms part of an antibody-stabilising  
moiety.

25 Compounds according to the invention may be per se  
pharmacologically active as well as linkable to monoclonal

antibodies.

Preferably the co-ordination compound of platinum has the general formula

5



10 in which A and B are the same or different selected from the class consisting of ammine and monodentate amine or A- and B together comprise a bidentate amine,

D is selected from the class consisting of substituted methylene and substituted dimethylene,

15 R is selected from the class consisting of hydrogen, lower (for example up to four carbon atoms) alkyl, aryl, aralkyl, alkenyl, cycloalkyl, cycloalkenyl, alkoxy and hydroxy,

X is a functionalised polymethylene moiety in which the functionalising group is selected from the class consisting of

20 carboxylic acid, alcohol, thiol or amine,

y is 1 if D is substituted methylene and is 1, 2 or 3 if D is substituted dimethylene and

z is 1 if D is substituted methylene and is 1 or 2 if D is substituted dimethylene,

25 with the proviso that where A and/or B is a monodentate cyclic alkylamine X may be a functionalising group selected from

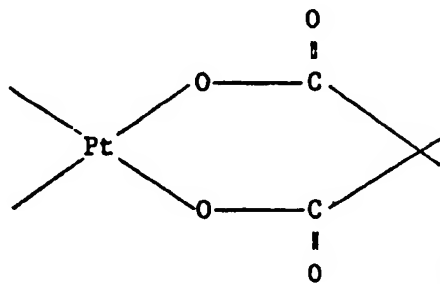
carboxylic acid, hydroxy, thio and amino.

The polymethylene chain X may optionally include ether, ester and/or peptide groups. Preferably the A and B groups are both cyclic alkylamine or both ammine.

Compounds according to the invention may exist as salts, for example as the potassium salt, or as solvated species.

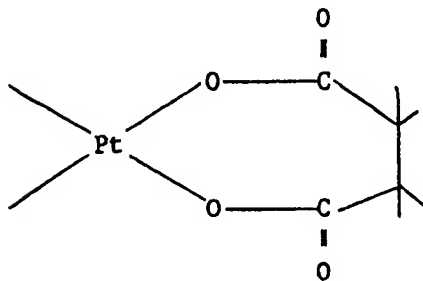
The antibody-stabilising moiety in the above formula is the six- or seven-membered ring, depending on whether D is substituted methylene or substituted dimethylene as follows:

15



six-membered ring

20

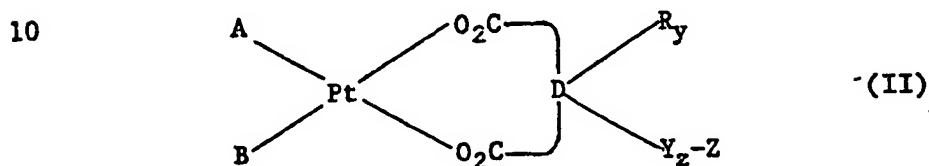


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seven-membered ring

The invention also includes a conjugate platinum co-ordination compound/monoclonal antibody complex in which the monoclonal antibody is linked to the platinum compound via a functional group which forms part of an in vivo antibody-stabilising moiety.

Conjugate complexes according to the invention have the general formula

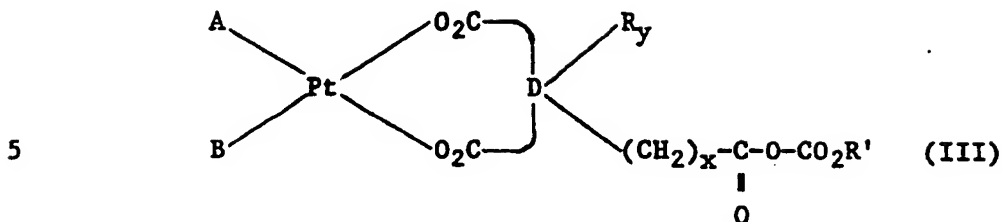


in which A, B, D, R, y and z have the same definition as in formula I, Y is a functionalised polymethylene moiety in which the functionalising group is selected from  $-\text{CO}_2-$ ,  $-\text{O}-$ ,  $-\text{S}-$  or  $-\text{NH}-$ , and Z is a monoclonal antibody.

The functionalised polymethylene moiety may optionally include ether, ester and/or peptide groups.

Preferably the  $-\text{Y}-\text{Z}$  moiety in the above formula II comprises or includes a peptide linkage although dioxide or disulphide bridge linkages are possible. In the formation of peptide-linked conjugate complexes, intermediate compounds comprising mixed anhydrides may be formed. Such intermediates

have the general formula



in which A, B, D, R, and y have the same definition as in formula I,  $\text{-(CH}_2\text{)}_x\text{-}$  is a polymethylene moiety optionally including  
 10 ether, ester and/or peptide groups, x is an integer (for example 1 to 15) and  
 R' is selected from the class consisting of straight chain, branched chain and cyclic alkyl groups.

15 It is also an aspect of this invention to provide intermediates having general formula III as defined above.

Referring to the general formula (I) for the co-ordination compound of platinum, the polymethylene group of the  
 20 functionalised polymethylene moiety X may have a total carbon chain length of C<sub>2</sub> to C<sub>20</sub>, preferably C<sub>2</sub> to C<sub>15</sub>, and may optionally include ether, ester and/or peptide groups. Examples of functionalised polymethylene moieties including ester groups are the short- and medium-chain polymethylene mono-esters of  
 25 dibasic carboxylic acids such as succinnic acid; examples of the



inclusion of ether groups are glycols.

Particular exemplary functionalised polymethylene moieties X optionally including ether and/or ester groups include  
5 the following:-

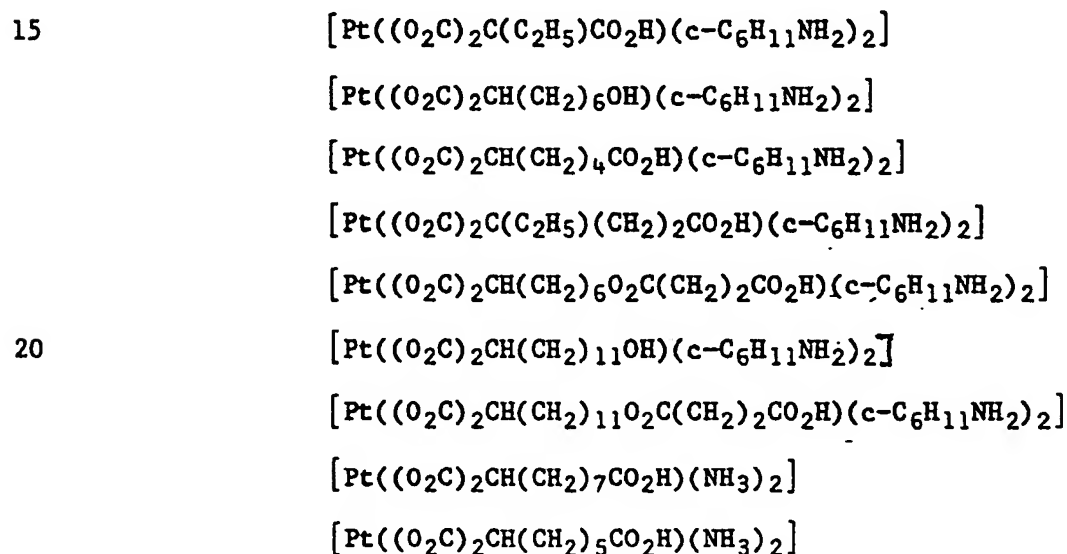
- $(\text{CH}_2)_2\text{CO}_2\text{H}$
- $(\text{CH}_2)_4\text{CO}_2\text{H}$
- $(\text{CH}_2)_5\text{CO}_2\text{H}$
- $(\text{CH}_2)_6\text{OH}$
- 10 - $(\text{CH}_2)_{10}\text{CO}_2\text{H}$
- $(\text{CH}_2)_{11}\text{OH}$
- $(\text{CH}_2)_2\text{NH}_2$
- $(\text{CH}_2\text{CH}_2\text{O})_3\text{H}$
- $\text{OCO}(\text{CH}_2)_2\text{CO}_2\text{H}$
- 15 - $(\text{CH}_2)_6\text{OCO}(\text{CH}_2)_2\text{CO}_2\text{H}$  and
- $(\text{CH}_2)_{11}\text{OCO}(\text{CH}_2)_2\text{CO}_2\text{H}$

Functionalised polymethylene moieties such as these are substituted for an acidic methylenic proton on a malonate or  
20 succinnate residue which forms, with platinum, an antibody-stabilising moiety comprising a six- or seven-membered ring. The other acidic methylenic proton (or protons) preferably remains (or remain) as such but may be substituted with lower alkyl, aryl, aralkyl, alkenyl, cycloalkyl, cycloalkenyl, alkoxy or  
25 hydroxy, as represented by R in formula (I). Where R is other than hydrogen, we prefer that it is lower alkyl, for example

methyl, ethyl or i-propyl.

The A and B groups are the same or different and are selected from ammine and monodentate amine or together represent bidentate amine. By "monodentate amine" is meant either a straight chain, branched chain or cyclic alkylamine or an arylamine or aralkylamine. By "bidentate amine" is meant a diamine such as ethylene diamine or an alicyclic or aromatic analogue thereof such as 1,2-diaminocyclohexane or 1,2-diaminobenzene.

Accordingly, exemplary co-ordination compounds of platinum according to the invention include



The invention also includes a pharmaceutical composition for the treatment of cancer and comprising an effective amount of

a compound of formula (I) or formula (II) in association with a pharmaceutically-acceptable carrier, diluent or excipient. Such compositions may be suitable for oral or parenteral administration, and may be in unit dosage form.

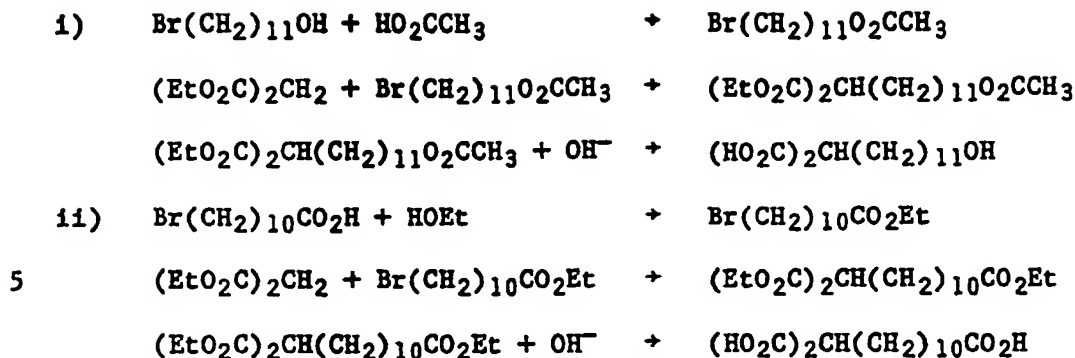
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Compounds according to the invention may in general be prepared by reacting a platinum compound having the formula  $\text{cis-}[\text{Pt(A)(B)I}_2]$  with the appropriate functionalised malonate or succinnate. Compounds having the formula  $\text{cis-}[\text{Pt(A)(B)I}_2]$  are  
10 prepared by the method of S.C. Dhara, Indian Journal of Chemistry, volume 8, page 193 (1970), the contents of which are herein incorporated by reference. The reaction with malonate or succinnate comprises reacting the platinum compound with aqueous silver nitrate to form the diaquo complex to which is then added  
15 the malonate or succinnate. The resulting precipitate is filtered off, washed and dried.

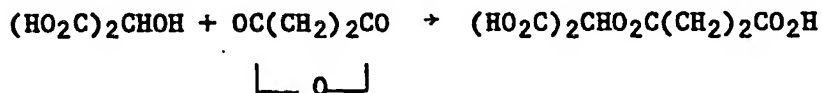
The functionalised malonate or succinnate may be prepared by taking an appropriate halide compound of the  
20 functionalising group and protecting the functionalising group, reacting the protected compound with diethyl malonate, and hydrolysing to remove the protective group so forming the acid from the diethyl ester.

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Examples of this reaction scheme are as follows:-

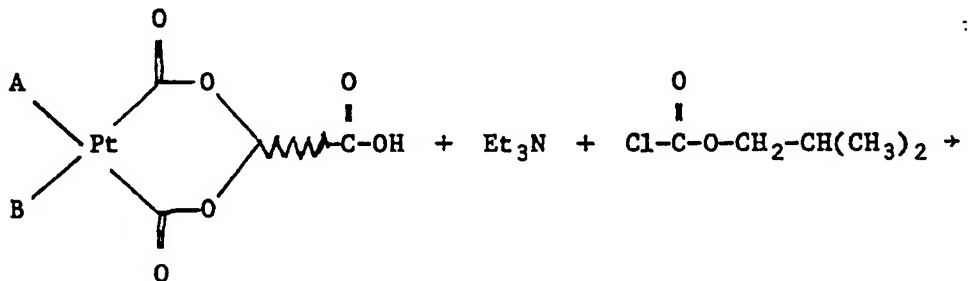


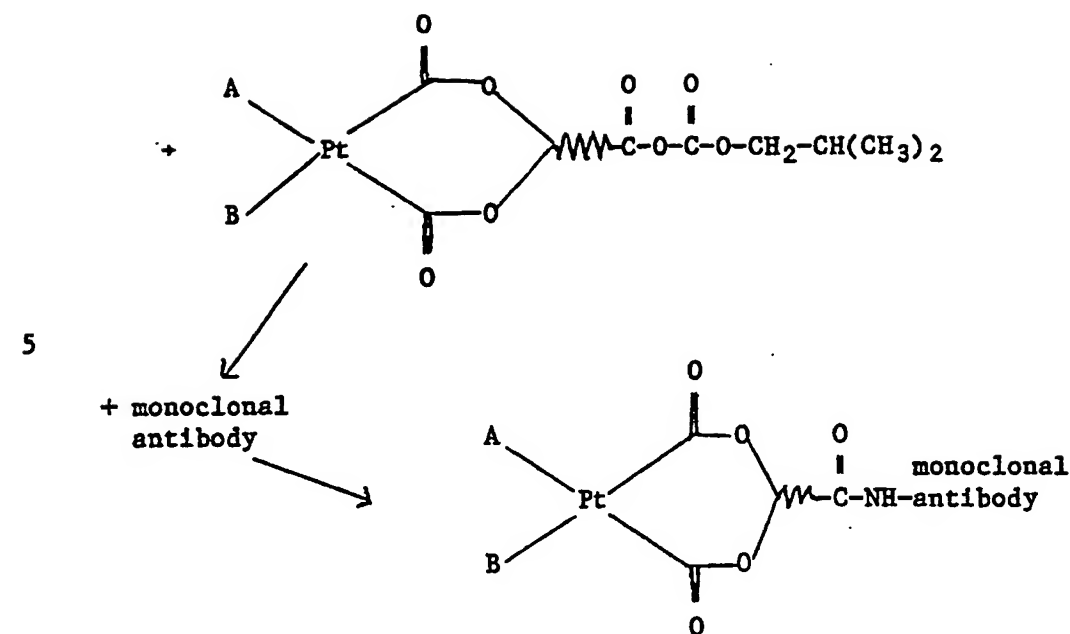
Where the functional group comprises an ester of a dibasic acid, the functionalised malonate may be prepared by refluxing tartronic acid with the dibasic acid anhydride in pyridine, for example



Conjugate platinum coordination compound/ monoclonal antibody complexes according to the invention may be prepared for example according to the following scheme, in which the functionalising group is  $-\text{CO}_2\text{H}$ :

20





According to the above scheme, the antibody is linked to the platinum compound via a peptide group, and the six-membered ring system imparts in vivo stability to the complex so that it reaches the desired site without having been hydrolysed. Once at the desired site, enzymes which are present in tumour cells are able to catalyse hydrolysis thereby releasing the active anti-tumour moiety. Hence the stability imparted should be sufficient to enable the complex to reach the desired site unattacked but should not be so great that the complex is not susceptible to hydrolysis at the desired site.

The preparation of various compounds according to the invention will now be described by way of example.

EXAMPLE 1

Preparation of  $[\text{Pt}\{(\text{O}_2\text{C})_2\text{CH}(\text{CH}_2)_4\text{CO}_2\text{H}\}(\text{c}-\text{C}_6\text{H}_{11}\text{NH}_2)_2]$

5                       $\text{Cis}-[\text{PtI}_2(\text{c}-\text{C}_6\text{H}_{11}\text{NH}_2)_2]$  was first prepared by the following method :

$\text{K}_2\text{PtCl}_4$  (50g, 0.12 mol) was dissolved in water (400mls). Charcoal (2g) was added and after stirring for two minutes the solution was  
10 filtered. An aqueous solution of KI (88.66g, 0.528 mol) was added to the stirred filtrate. Cyclohexylamine (29.7g, 0.3 mol) was added to the mixture which was stirred for a further 30 minutes at room temperature. The mixture was filtered and the precipitate was washed with water ( 3 x 150ml), ethanol (2 x 150ml) and  
15 diethylether (1 x 150ml). The product was dried in air overnight at 50°C. Yield 73.7g (94.1%).

The compound according to the invention was then prepared as follows :-

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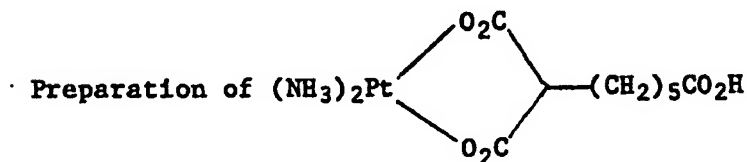
$[\text{PtI}_2(\text{c}-\text{C}_6\text{H}_{11}\text{NH}_2)_2]$  (5g, 0.0077 mol) was added to a stirred aqueous solution of  $\text{AgNO}_3$  (2.6g, 0.015 mol). The solution was heated with stirring at 50°C for 3h and filtered. A solution of  $(\text{HO}_2\text{C})_2\text{CH}(\text{CH}_2)_4\text{CO}_2\text{H}$  (1.88g, 9.2 mmol) was partially neutralised  
25 with KOH (0.168g, 3 mmol) and added to the filtrate above. The

mixture was stirred overnight at room temperature and filtered to give a cream precipitate which was washed with water, ethanol and diethylether before being dried overnight. Yield (2.2g, 48%).

5	<u>Elemental analysis</u>	C	H	N
	calculated(as monohydrate)	39.15%	6.24%	4.56%
	found	40.39%	5.93%	5.44%

## EXAMPLE 2

10



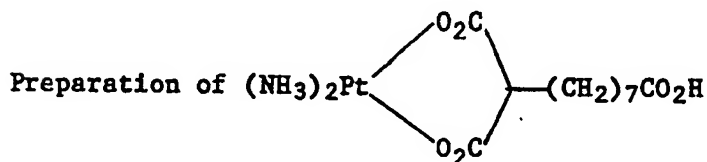
- 15                   Cis-diammine-diiodo-platinum(II) (3.65g, 7.6 mmol) was added as a solid to silver nitrate (2.52g, 4.8 mmol) in water (30ml) and stirred in the dark for 2 $\frac{1}{2}$  hours. Silver iodide was filtered as a yellowish-brown solid. The sodium chloride test for silver ion was negative for the yellow filtrate.
- 20 2-Carboxyoctanedioic acid, (1.66g, 7.6 mmol) was heated with sodium hydroxide (0.91g, 22.9 mmol) dissolved in 30ml water. The clear solution was brought to pH 6 with concentrated nitric acid. The 2-carboxyoctanedioic acid and cis-diammine-diaquoplatinum(II) solutions were combined with stirring at 50°C for two hours. The
- 25 clear pale yellow solution was acidified to pH 4 with concentrated nitric acid, affording a pale green precipitate. After standing

18 hours, the green product was washed with water, ethanol and ether and dried in vacuo (2.24g, 72.2% yield).

The 2-carboxyoctanedioic malonate complex was recrystallized with a large volume of hot water, concentrated and chilled for 18 hours. A white solid was collected. Infrared (KBr) 3250 to 3100, 2930, 1700 to 1550  $\text{cm}^{-1}$ .

### EXAMPLE 3

10



15

The method of Example 2 was followed, using the 2-carboxydecanedioic acid ligand (1.87g, 7.6 mmol). The product was recrystallized from acetone/water to afford a white solid (2.14g, 66.1%), Infrared (KBr) 3250 to 3090, 2920, 2850, 1760 to 1550  $\text{cm}^{-1}$ .

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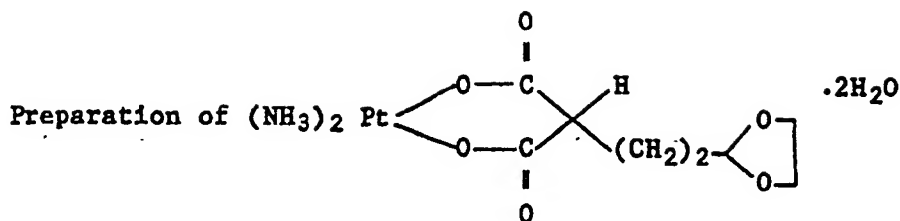
<u>Elemental analysis</u>	C	H	N
calculated	27.90	4.69	5.92
found	27.45	4.54	5.88.

25

The following compounds were also prepared by the method of either Example 1 or Example 2 :



<u>COMPOUND</u>		<u>ANALYSIS</u>		
		C	H	N
	$[\text{Pt}\{(\text{O}_2\text{C})_2\text{C}(\text{C}_2\text{H}_5)\text{CO}_2\text{H}\}(\underline{\text{c}}\text{-C}_6\text{H}_{11}\text{NH}_2)]$			
	found %	38.84	6.06	4.89
5	(As $\text{H}_2\text{O}$ adduct) calc. %	39.15	6.24	4.56
	$[\text{Pt}\{(\text{O}_2\text{C})_2\text{CH}(\text{CH}_2)_4\text{CO}_2\text{H}\}(\underline{\text{c}}\text{-C}_6\text{H}_{11}\text{NH}_2)_2]$			
	found %	40.39	5.93	5.44
	(As $\text{H}_2\text{O}$ adduct) calc. %	39.15	6.24	4.56
10	$[\text{Pt}\{(\text{O}_2\text{C})_2\text{CH}(\text{CH}_2)_6\text{OK}\}(\underline{\text{c}}\text{-C}_6\text{H}_{11}\text{NH}_2)_2]$			
	found %	38.57	6.69	4.78
	as monohydrate calc. %	38.70	6.34	4.30
15	$[\text{Pt}\{(\text{O}_2\text{C})_2\text{CH}(\text{CH}_2)_{11}\text{OK}\}(\underline{\text{c}}\text{-C}_6\text{H}_{11}\text{NH}_2)_2]$			
	found %	42.92	7.28	3.58
	as monohydrate calc. %	43.26	7.12	3.88
	$[\text{Pt}\{(\text{O}_2\text{C})_2\text{CH}(\text{CH}_2)_{11}\text{O}_2\text{CCH}_2\text{CH}_2\text{CO}_2\text{H}\}(\underline{\text{c}}\text{-C}_6\text{H}_{11}\text{NH}_2)_2]$			
20	found %	47.34	7.76	3.57
	calc. %	47.05	7.11	3.66
	$[\text{Pt}\{(\text{O}_2\text{C})_2\text{CH}(\text{CH}_2)_7\text{CO}_2\text{H}\}(\text{NH}_3)_2]$			
	found %	27.45	4.54	5.88
25	calc. %	27.90	4.69	5.92

EXAMPLE 4

3.0g of  $\text{cis-}[\text{PtI}_2(\text{NH}_3)_2]$  (6.2 mmol) was reacted with  $\text{AgNO}_3$  (2.07g, 12%) in water (40ml) to form the diaquo species.

This solution was added dropwise to a solution of 2-[2-(1,3-dioxolanyl)]ethyl propanedioic acid (1.3g, 6.2 mmol). The pH of the reaction mixture was maintained at 6.05 using bicarbonate solution. The clear solution was heated to  $60^\circ\text{C}$  for 2h and stirred at room temperature for 70h. The reaction mixture was evaporated to dryness and redissolved in a minimal volume of water. Chilling this solution yielded a white, crystalline solid, which was collected and washed with EtOH and Et<sub>2</sub>O. Yield = 0.91g, 31% yield.

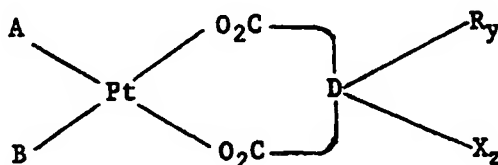
Elemental analysis	C	H	N
calculated ( $\cdot 2\text{H}_2\text{O}$ )	20.58	4.32	6.00
found	19.45	3.98	6.18

CLAIMS

1. A co-ordination compound of platinum linkable to a monoclonal antibody, wherein the compound includes an antibody-linkable functional group which forms part of an antibody-stabilising moiety.

2. A co-ordination compound of platinum according to claim 1 having the general formula

10



15 in which A and B are the same or different selected from the class consisting of ammine and monodentate amine or A and B together comprise a bidentate amine,

D is selected from the class consisting of substituted methylene and substituted dimethylene,

20 R is selected from the class consisting of hydrogen, lower alkyl, aryl, aralkyl, alkenyl, cycloalkyl, cycloalkenyl, alkoxy and hydroxy,

X is a functionalised polymethylene moiety in which the functionalising group is selected from the class consisting of  
25 carboxylic acid, alcohol, thiol or amine,

y is 1 if D is substituted methylene and is 1 or 2 if D is substituted dimethylene and

z is 1 if D is substituted methylene and is 1 or 2 if D is substituted dimethylene,

with the proviso that where A and/or B is a monodentate cyclic alkylamine X may be a functionalising group selected from carboxylic acid, hydroxy, thio and amino.

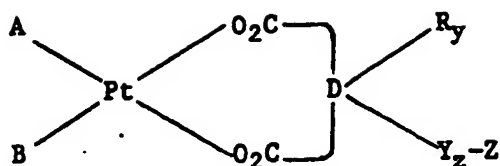
3. A compound according to claim 2 in which the polymethylene chain X included ether, ester and/or peptide groups.

4. A compound according to claim 2 in which the A and B groups are both cyclic alkylamine or both ammine.

5. A compound according to claim 2 or claim 3 in which the functionalised polymethylene moiety X has a total carbon chain length of C<sub>2</sub> to C<sub>20</sub>.

6. A conjugate platinum co-ordination compound/monoclonal antibody complex in which the monoclonal antibody is linked to the platinum compound via a functional group which forms part of an in vivo antibody-stabilising moiety.

7. A conjugate complex according to claim 6 having the general formula

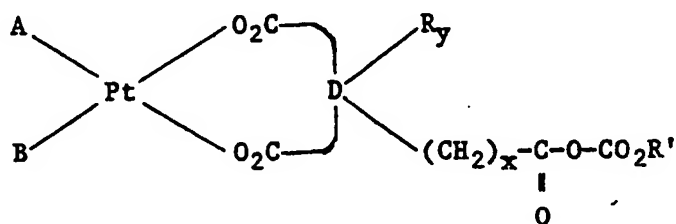


5 in which A, B, D, R, y and z have the meanings defined in claim 2, Y is a functionalised polymethylene moiety in which the functionalising group is selected from the class consisting of  $\text{-CO}_2\text{-}$ ,  $\text{-O-}$ ,  $\text{-S-}$  and  $\text{-NH-}$ , and Z is a monoclonal antibody.

10 8. A conjugate complex according to claim 7 in which the functionalised polymethylene moiety includes ether, ester and/or peptide groups.

9. Mixed anhydride compounds having the general formula

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in which A, B, D and R and y have meanings as defined in claim 2,  $\text{-(CH}_2\text{)}_x\text{-}$  is a polymethylene moiety optionally including ether, ester and/or peptide groups and x is an integer from 1 to 15 and R' is selected from the class consisting of straight chain, 25 branched chain and cyclic alkyl groups.

10. A pharmaceutical composition for the treatment of cancer and comprising an effective amount of a compound according to any of claims 1 to 9 in association with a pharmaceutically-acceptable carrier, diluent or excipient.
11. A composition according to claim 10 in unit dosage form.



DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl. 4)
A	US-A-4 115 418 (G.R.GALE et al.) * Claims 1,2 *	1,2,4	C 07 F 15/00 C 07 K 15/00 A 61 K 31/28 A 61 K 39/395
A	FR-A-2 401 933 (THE UNITED STATES GOVERNMENT) * Claim 1 *	1,2,4	
A	US-A-4 140 707 (M.J.CLEARE et al.) * Claims 1-4; column 2, lines 3-21 *	1,2,4	
A	GB-A-2 006 776 (JOHNSON, MATTHEY & CO.) * Claims 1,3,5-9 *	1,2,4	
A	US-A-4 169 846 (Y.KIDANI et al.) * Column 4, line 6 - column 5, line 38 *	1,2,4	TECHNICAL FIELDS SEARCHED (Int. Cl. 4)  C 07 F A 61 K
D,A	EP-A-0 099 133 (YEDA RESEARCH AND DEVELOPMENT CO.) * Claims; page 3, lines 25-30 *	1,6,7,10	
The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 04-10-1985	Examiner RYCKEBOSCH A.O.A.
<b>CATEGORY OF CITED DOCUMENTS</b>			
X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document	